

OVERVIEW		
Full Name	Copper (Cuprum)	
Symbol	Cu	
Atomic Number	29	
Series	Transition Metal	
Block / Group / Period	d-block   Group 11   Period ...	
Electron Config.	[Ar]4s13d10	
Phase at STP	Solid	
Colour	Copper	
Discovery	8000 BC	
THERMAL PROPERTIES		
Melting Point	1084.62 °C	°C
Melting Point (abs.)	1357.77 K	K
Boiling Point	2562 °C	°C
Boiling Point (abs.)	2835 K	K
Heat of Fusion	13.1 kJ/mol	kJ/mol
Heat of Vaporization	300 kJ/mol	kJ/mol
Specific Heat (solid)	384.4 J/(kg K)	J/(kg·K)
Thermal Conductivity	400 W/(m K)	W/(m·K)
Thermal Expansion	0.0000165 K <sup>-1</sup>	K <sup>-1</sup>
MECHANICAL PROPERTIES		
Density (solid, STP)	8.96 g/cm <sup>3</sup>	g/cm <sup>3</sup>
Density (liquid, MP)	8.02 g/cm <sup>3</sup>	g/cm <sup>3</sup>
Young's Modulus	130 GPa	GPa
Shear Modulus	48 GPa	GPa
Bulk Modulus	140 GPa	GPa
Poisson's Ratio	0.34	
Speed of Sound	3570 m/s	m/s
Mohs Hardness	3	
Vickers Hardness	369 MPa	MPa
Brinell Hardness	874 MPa	MPa
ELECTRICAL PROPERTIES		
Type	Conductor	
Conductivity	5.9×10 <sup>7</sup> S/m	S/m
Resistivity	1.7×10 <sup>-8</sup> m ?	Ohm·m
MAGNETIC PROPERTIES		
Type	Diamagnetic	
Vol. Susceptibility	-9.63×10 <sup>-6</sup>	
Mass Susceptibility	-1.08×10 <sup>-9</sup> m <sup>3</sup> /Kg	m <sup>3</sup> /kg
REACTIVITY		
Valence	2	
Electronegativity	1.9	(Pauling)
Electron Affinity	118.4 kJ/mol	kJ/mol
Ionization Energies	745.5, 1957.9	kJ/mol
ATOMIC DIMENSIONS		
Atomic Radius	145 pm	pm
Covalent Radius	132 pm	pm
Van der Waals Radius	140 pm	pm
CRYSTAL STRUCTURE		
Structure	Face-centered Cubic	
Lattice Constant	361.49, 361.49, 361.49 pm	
Space Group	Fm_ 3m	#225
ISOTOPES & NUCLEAR		
Stability	Stable	
Stable Isotopes	63Cu, 65Cu	
Isotopic Abundance	63Cu69.17%65Cu30.83%	
Known Isotopes	52Cu ■ 80Cu (29 total)	
Neutron Cross Sect.	3.78	barns
ELEMENTAL ABUNDANCES		
Universe	6×10 <sup>-6</sup> %	%
Earth's Crust	0.0068%	%
Oceans	3×10 <sup>-7</sup> %	%
Human Body	0.0001%	%
HEALTH & SAFETY		
DOT Hazard Class	4.1	
RTECS Number	GL5325000	
NFPA Health/Fire/Rx	0 / 0 / 0	